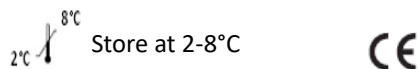


**Chloride**  
**Thiocyanate-Hg. Colorimetric**  
**Quantitative determination of chloride ion**

**IVD** For in-vitro diagnostic and professional use only.



**INTENDED USE**

For the measurement of chloride ion concentration in human body fluids.

**INTRODUCTION**

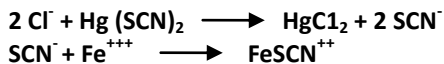
It is important clinically the determination of chloride due regulation of osmotic pressure of extra cellular fluid and to its significant role in acid-base balance. Increases in chloride ion concentration may be found in severe dehydration, excessive intake of chloride, severe renal tubular damage and in patients with cystic fibrosis.

Decrease in chloride ion concentration may be found in metabolic acidosis, loss from prolonged vomiting and chronic pyelonephritis.

Clinical diagnosis should not be made on a single test result; it should integrate clinical and other laboratory data.

**PRINCIPLE**

The quantitative displacement of thiocyanate by chloride from mercuric thiocyanate and subsequent formation of a red ferric thiocyanate complex is measured calorimetrically.



The intensity of the color formed is proportional to the chloride ion concentration in the sample.

**REAGENTS**

|  |                      |           |
|--|----------------------|-----------|
| <b>R</b>                                       | Mercuric thiocyanate | 4 mmol/L  |
| <b>Mercuric-Thiocyanate (Chloride Reagent)</b> | Ferric nitrate       | 40 mmol/L |
|  | Mercuric nitrate     | 2 mmol/L  |
|  | Nitric acid          | 45 mmol/L |

|                          |   |
|--------------------------|---|
| <b>Chloride standard</b> | Chloride aqueous primary standard<br>125 mmol/L |
|--------------------------|---|

**PRECAUTIONS**

- Mercury(II) thiocyanate:Harmful(Xn):Hamrful by inhalation, in contact with skin and if sw allowed.
- Keep away from food, drink and animal feeding stuffs.
- After contact with skin, wash immediately with plenty of water.
- In case of accident or if you feel unwell, seek medical advice immediately (show the label where possible).
- This material and its container must be disposed of as hazardous.
- Avoid release to the environment Refer to special instructions/ safety data sheets.

**PREPARATION:**

- All the reagents are ready to use.

**STORAGE AND STABILITY**

- All the components of the kit are stable until the expiration date on the label when stored tightly closed at 2-8°C, protected from light and contaminations prevented during their use.
- Do not use reagents over the expiration date.
- Signs of reagent deterioration:
  - Presence of particles and turbidity
  - Blank absorbance (A) at 480 nm ≥0.15.

**EQUIPMENTS NEEDED BUT NOT PROVIDED**

- Spectrophotometer or colorimeter measuring at 480 nm.
- Matched cuvettes 1.0 cm light Path.
- General laboratory equipment.

**SAMPLES**

- Serum, plasma: Free of hemolysis and separated from cells as rapidly as possible. Anticoagulants such as oxalate or EDTA are not acceptable they will interfere with results.
- Urine: Collect 24-hour urine specimen in chloride

free containers. Dilute a sample 1/2 in distilled water. Mix Multiply results by 2 (dilution factor).

- Stability of the sample: Ion chloride is stable 1 week at room temperature (15-25°C), in refrigerator (2-8°C) or frozen (-20°C) temperatures.

**PROCEDURE**

- Assay conditions:  
Wavelength:.....480 (440-500) nm  
Cuvette:.....1 cm. light path  
Temperature:.....37°C /15-25°C
- Adjust the instrument to zero with distilled water.
- Pipette into a cuvette:

|                      | Blank | Standard | Sample |
|----------------------|-------|----------|--------|
| <b>R (ml)</b>        | 1.0   | 1.0      | 1.0    |
| <b>Standard (µl)</b> | --    | 10       | --     |
| <b>Sample (µl)</b>   | --    | --       | 10     |

- Mix and incubate for 5 min at 37°C /15-25°C.
- Read the absorbance (A) of the samples and Standard, against the Blank the color is stable for at least 30 minutes.

**CALCULATIONS**

$\frac{(A) \text{ Sample} - (A) \text{ Blank}}{(A) \text{ Standard} - (A) \text{ Blank}} \times 125 \text{ (Stand. conc.)} = \text{mmol/L (chloride in the sample)}$

Urine 24 h:  
 $\frac{(A) \text{ Sample} - (A) \text{ Blank}}{(A) \text{ Standard} - (A) \text{ Blank}} \times 125 \times \text{vol. (dL) urine/24h} = \text{mmol/24 h (chloride in the sample)}$

Conversion factor: mmol/L = mEq/L.

**QUALITY CONTROL**

Control sera are recommended to monitor the performance of assay procedures.

If control values are found outside the defined range, check the instrument, reagents and calibrator for problems.

Each laboratory should establish its own Quality Control scheme and corrective actions if controls do not meet the acceptable tolerances.

## REFERENCE VALUES

|                 |                   |
|-----------------|-------------------|
| Serum or plasma | 95-115 mmol/L     |
| Urine           | 110-250 mmol/24 h |

These values are for orientation purpose; each laboratory should establish its own reference range.

## PERFORMANCE CHARACTERISTICS

### Measuring range:

From *detection limit* of 0.454 mmol/L to *linearity limit* of 190 mmol/L.

If the results obtained were greater than linearity limit, dilute the sample 1/2 with distilled water and multiply the result by 2.

### Precision:

|               | Intra-assay (n=20) |      | Inter-assay (n=20) |      |
|---------------|--------------------|------|--------------------|------|
|               | Mean (mmol/L)      | SD   | Mean (mmol/L)      | SD   |
| Mean (mmol/L) | 84.2               | 0.81 | 114                | 0.62 |
| SD            | 0.81               | 0.62 | 82.5               | 1.07 |
| CV %          | 0.96               | 0.55 | 1.30               | 1.68 |

### Sensitivity.

1 mmol/L = 0.00471 A.

### Accuracy:

Results obtained using reagents did not show systematic differences when compared with other commercial reagents. The results obtained using 50 samples were the following:

Correlation coefficient ( $r^2$ ): 0.96731

Regression equation:  $y=0.990x+0.100$

The results of the performance characteristics depend on the analyzer used.

## INTERFERENCES

- Hemolysis. Anticoagulants other than heparin.
- Bilirubin up to 120 mg/L, bovine serum albumin up to 150 g/L and triglycerides up to 6 g/L did not significantly alter the assay.
- A list of drugs and other interfering substances with chloride determination has been reported by Young.

## NOTES


1. It is recommended to use disposable material. If glassware is used the material should be

scrupulously cleaned with H<sub>2</sub>SO<sub>4</sub> - K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> Solution and then thoroughly rinsed it with distilled water.

2. Most of the detergents and water softening products used in the laboratories contains chelating agents. A defective rinsing will invalidate the procedure.
3. Avoid the contact with metal materials.
4. Calibration with the aqueous standard may cause a systematic error in automatic procedures. In these cases, it is recommended to use a serum Calibrator.
5. Use clean disposable pipette tips for its dispensation.
6. Chloride standard: proceed carefully with this product because due to its nature, easily it can get contaminated.

## REFERENCES

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









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|   |   |   |                                    |
|---|---|---|------------------------------------|
|  REF | Catalogue Number                                    |  | Temperature limit                  |
|  IVD | In Vitro diagnostic medical device                  |  | Caution                            |
|      | Contains sufficient for <n> tests and Relative size |  | Consult instructions for use (IFU) |
|  LOT | Batch code  |  | Manufacturer                       |
|      | Fragile, handle with care                           |  | Use-by date                        |
|      | Manufacturer fax number                             |  | Do not use if package is damaged   |
|      | Manufacturer telephone number                       |  | Date of Manufacture                |
|      | Keep away from sunlight                             |  | Keep dry                           |